



DRAFT:

Metal Oxide based Nanostructure for Multifunctional Devices Applications.

Yudi Darma

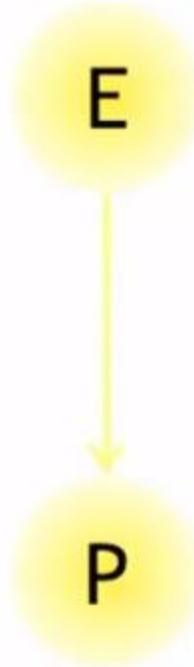
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Functionality: Magnetolectric Response

Controlling P and M with E and H fields

Usually:

Control electrical properties
with an electric field

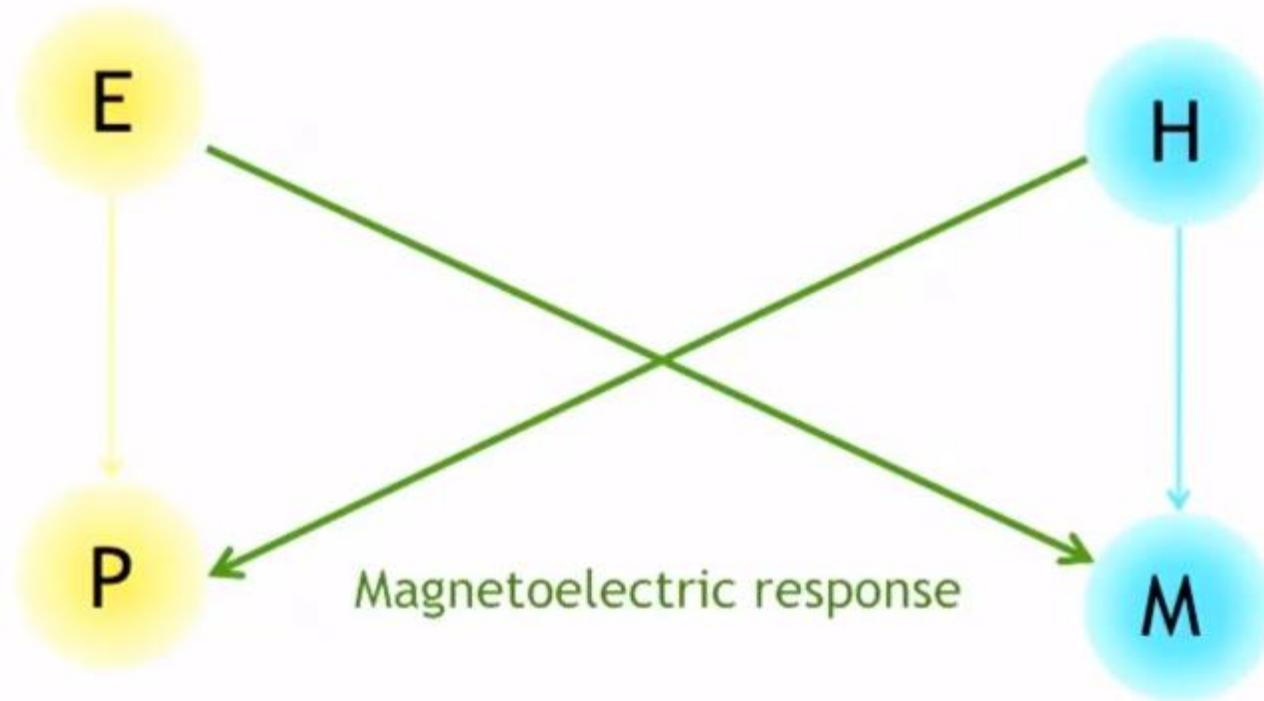


Control magnetic properties
with a magnetic field



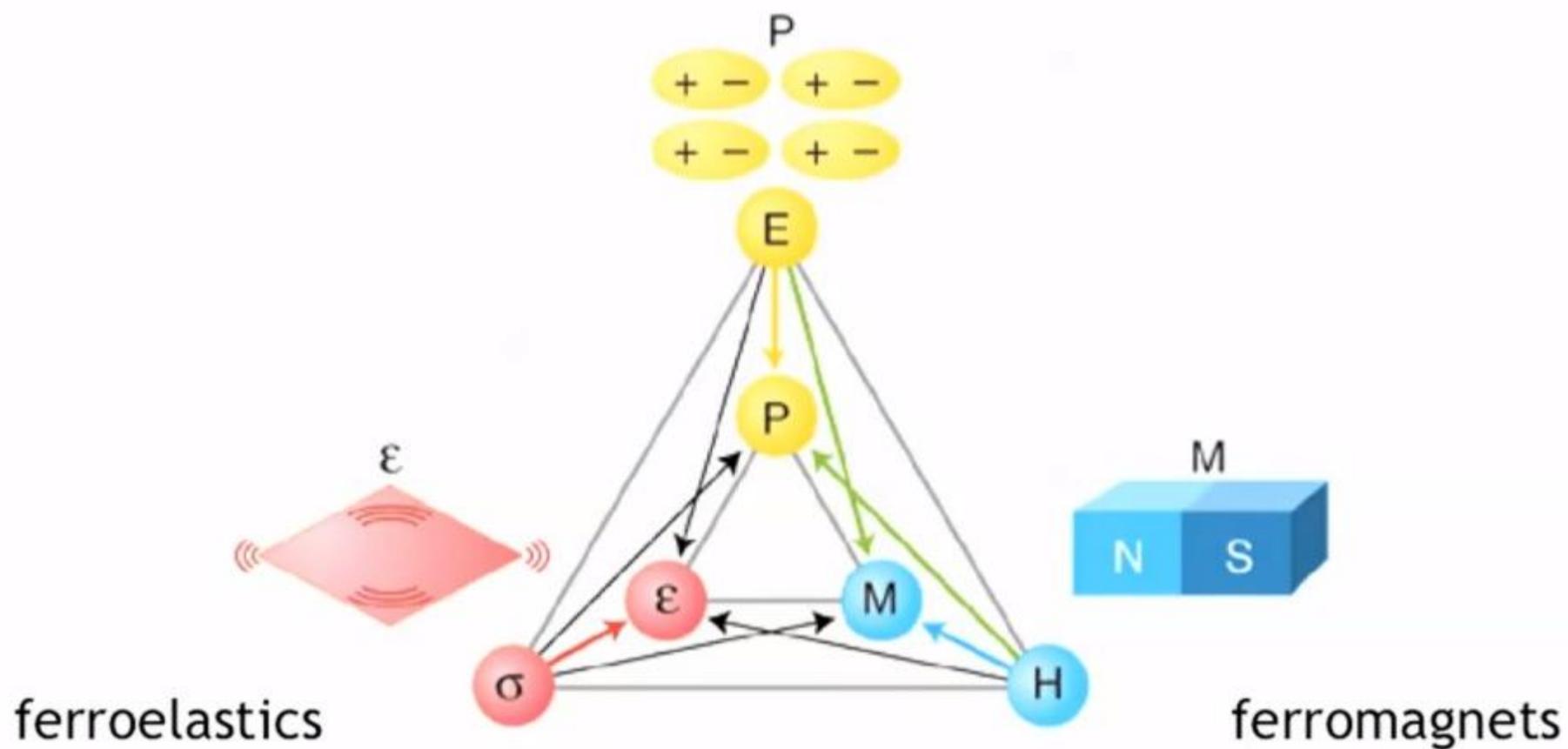
Functionality: Magnetolectric Response

Instead:



Multiferroics and Magnetoelectricity

ferroelectrics

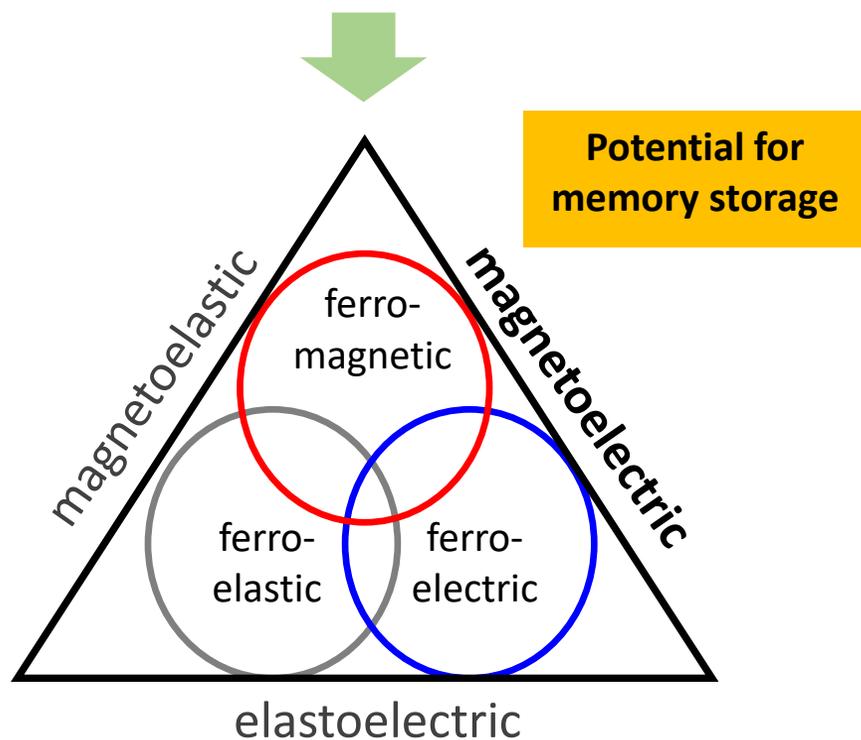


The renaissance of magnetoelectric multiferroics,
N. A. Spaldin and M. Fiebig, *Science* **15**, 5733 (2005)

Multiferroic: Potential of multifunctionality

Multiferroic (materials) characteristic → multifunctional devices

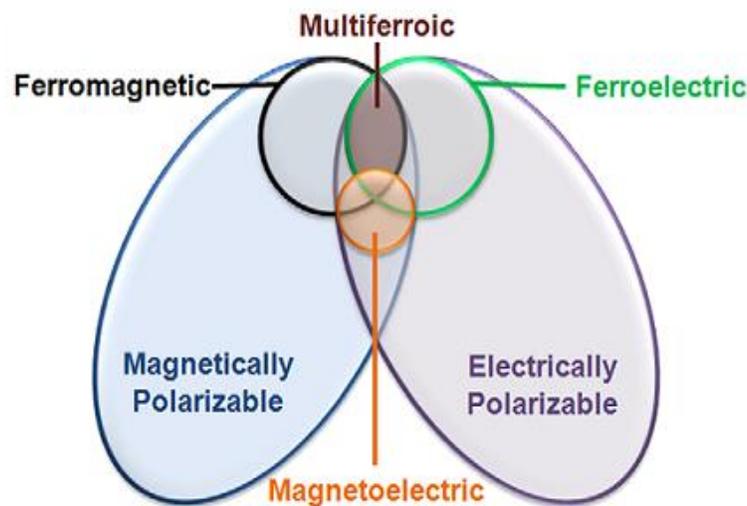
→ Simultaneous existence of two or more ferroic orders in one material system.



Relation between ferroelectric and ferromagnetic in multiferroic system

Ferroelectric (FE): electrically polarizable >< **Ferromagnetic (FM):** magnetically polarizable

→ Coexistence FE and FM in the systems generates multiferroic characteristic (magnetoelectric).



(Eerenstein *et al.*, *Nature*, 2006, 442, 759)

However,....

Challenge → to simultaneously produce ferroelectric and ferromagnetic

▪ Ferroelectric

- appears in material with **high crystallinity (low defects)**.
- possess ions that have a formal $3d^0$ electronic state.

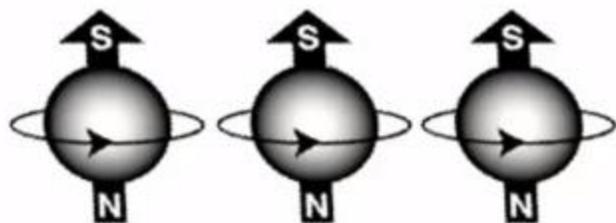
▪ Ferromagnetic

- appears in material with **low crystallinity (high defect)**.
- possess incompletely filled 3d shells.

Why are there so few magnetic ferroelectrics?

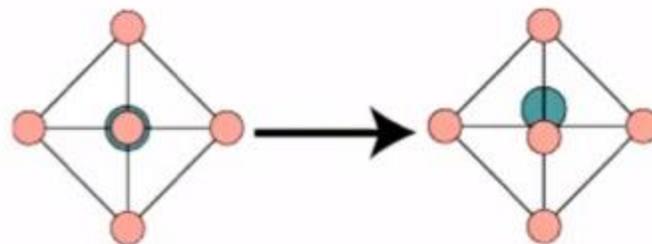
the chemistry that promotes one functionality often prohibits another

Ferromagnetism



Requires partially filled d orbitals
(Hund versus Pauli)

Ferroelectricity



Requires empty d orbitals; “Matthias rule”
(covalency versus Coulomb repulsion)

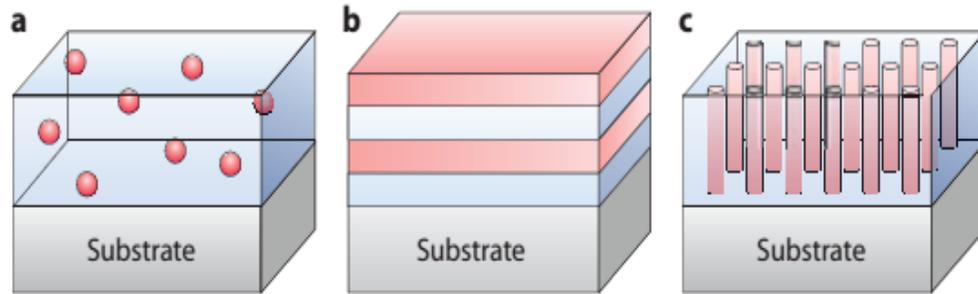
CHEMICALLY CONTRA-INDICATED!

B.T. Matthias, *New ferroelectric crystals*, Phys. Rev. (1949)
N.A. Hill, *Why are there so few magnetic ferroelectrics?* J. Phys. Chem. B 104, 6694 (2000)

Solution?

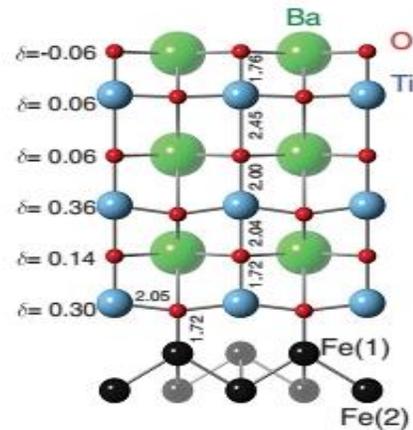
Ferroelectric and ferromagnetic materials is combined in several configuration.

(1) Composite



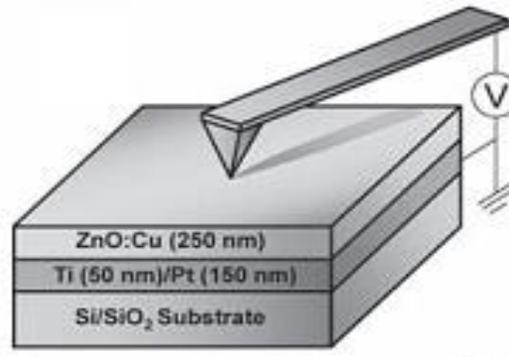
Y. Wang, et al. *NPG Asia Mater.*, 2010, 2(2) 61–68

(2) Interface effects



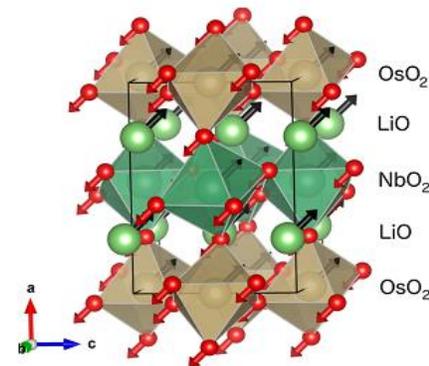
Meyerheim et al., *Phys. Rev. Lett.*, 2011, 106, 0872032011

(3) Doping



T. S. Heng et al., *Adv. Mater.*, 2011, 23(14), 1635-1640

(4) Superlattice



Puggioni et al., *Phys. Rev. Lett.*, 2015, 115, 087202

ZnO

Electronics,

Optic and

Magnetic

Room-Temperature Ferromagnetism of Cu-Doped ZnO Films Probed by Soft X-Ray Magnetic Circular Dichroism

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⁴*CMPMSD, Brookhaven National Laboratory, Upton, New York 11973, USA*

⁵*Department of Physics and Astronomy, SUNY Stony Brook, Stony Brook, New York 11794, USA*

⁶*School of Physics, Shandong University, Jinan 250100, People's Republic of China*

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⁸*National Synchrotron Light Source, Brookhaven National Laboratory, Upton, New York 11973, USA*

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(Received 24 November 2009; published 8 November 2010)

We report direct evidence of room-temperature ferromagnetic ordering in O-deficient ZnO:Cu films by using soft x-ray magnetic circular dichroism and x-ray absorption. Our measurements have revealed

- ✓ RT Ferromagnetism (FM) of 2% Cu doped ZnO
- ✓ Vo contribution and Cu impurities is essential to observe FM
- ✓ Large size vacancies orbital → by first principle calculation.



Mutual Ferromagnetic–Ferroelectric Coupling in Multiferroic Copper-Doped ZnO

Tun Seng Heng, Meng Fei Wong, Dongchen Qi, Jiabao Yi, Amit Kumar, Alicia Huang, Fransiska Cecilia Kartawidjaja, Serban Smadici, Peter Abbamonte, Cecilia Sánchez-Hanke, Santiranjan Shannigrahi, Jun Min Xue, John Wang, Yuan Ping Feng, Andrivo Rusydi, Kaiyang Zeng,* and Jun Ding**

There is tremendous flurry of research interest in multiferroic materials that exhibit multiple primary ferroic order parameters simultaneously and that have practical applications.^[1]

multiferroic material was hampered and was mainly based on silicon incompatible perovskite materials,^[1] which limits their multifunctional applications.

- ✓ Mutual ferroelectric and ferromagnetic in Cu Doped ZnO
- ✓ This behavior sensitive to Cu density and V_o
- ✓ Critical [Cu] is around 8%

Interplay of Cu and oxygen vacancy in optical transitions and screening of excitons in ZnO:Cu films

Yudi Darma,^{1,2,3,4} Tun Seng Heng,^{1,5} Resti Marlina,⁴ Resti Fauziah,⁴ Jun Ding,⁵ and Andrivo Rusydi^{1,2,3,4,a)}

¹Singapore Synchrotron Light Source, National University of Singapore, 5 Research Singapore

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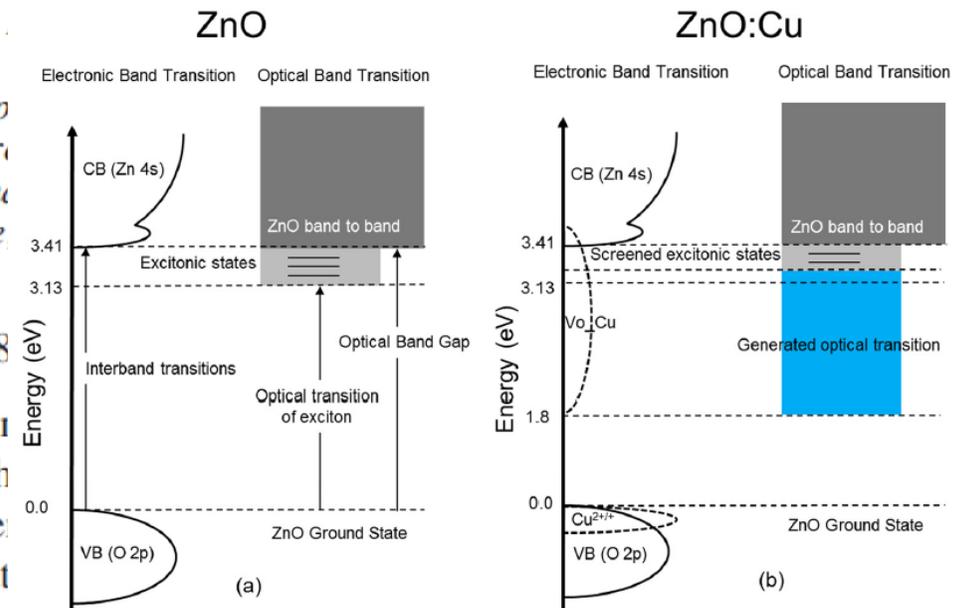
³NUSNNI-NanoCore, National University of Singapore, 2 Science Drive 3, Singapore

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⁵Department of Material Science and Engineering, National University of Singapore Singapore 117574, Singapore

(Received 1 December 2013; accepted 5 February 2014; published online 28 February 2014)

We study room temperature optics and electronic structures of ZnO:Cu film concentration using a combination of spectroscopic ellipsometry, photoluminescence, and ultraviolet-visible absorption spectroscopy. Mid-gap optical states, interband excitons are observed and distinguishable. We argue that the mid-gap states

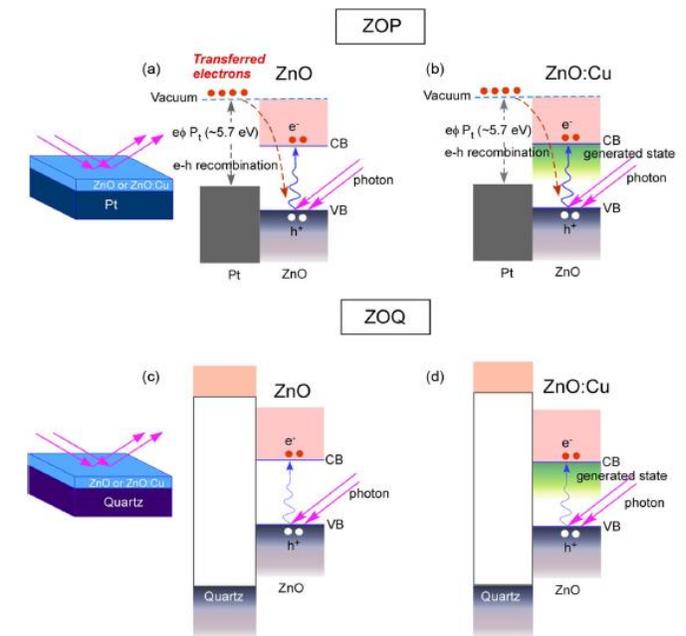


- ✓ Observation of exciton and new transition state
- ✓ Screening of exciton by transition states
- ✓ The role of Cu concentration to optical conductivity

Strong Modification of Excitons and Optical Conductivity for Different Dielectric Environments in ZnO Films

Volume 8, Number 3, June 2016

Yudi Darma
Resti Marlina
Tun Seng Herng
Jun Ding
Andrivo Rusydi



- **Electronic blocking** and **screening effects** in ZnO film through dielectric function and optical conductivity as a function of substrates and Cu concentration have been studied systematically
- Strong modification of excitonic states in ZnO films by selecting different electronic environments

Surface modification on ZnO based thin film

→ Multiple-stacked nano porous ZnO

Surface modification effect to **optical** properties

→ Emission enhancement in ZnO are still become the main interest in thin film for photonic/optoelectronics devices

→ ZnO Nanocolumnar (NC)

modification of **magnetic** and **electronic** properties

→ Reduction of ZnO dimension to increase Zn vacancy at the surface



ZnO Film → Multiple-stacked nano porous ZnO

Surface modification effect to optical properties

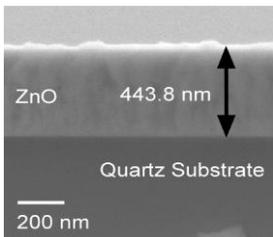
ZnO Green Emission Sample

PLD

Laser energy ~180mJ, frequency 5Hz

Sample 1

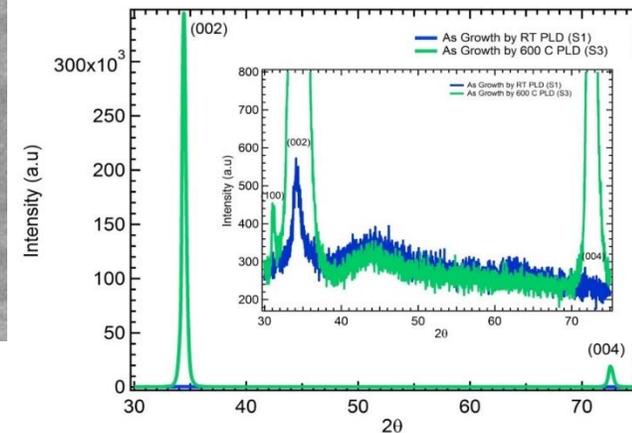
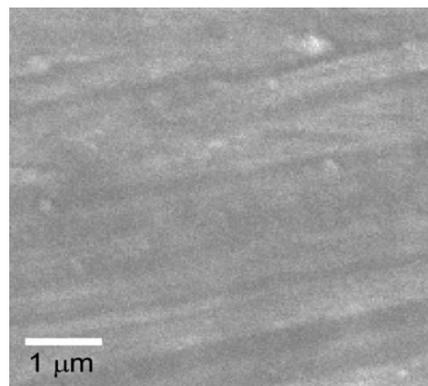
Sample 3



as prepare
Vacuum 10^{-7} mbarr
(no oxygen supply)
RT
~440nm



as prepare
 2×10^{-4} mbarr
(oxygen supply)
600 C
~720nm



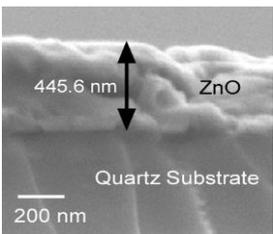
H Annealing

annealed in hydrogen (Ar 95%+H5%)
570 C for 1hour.

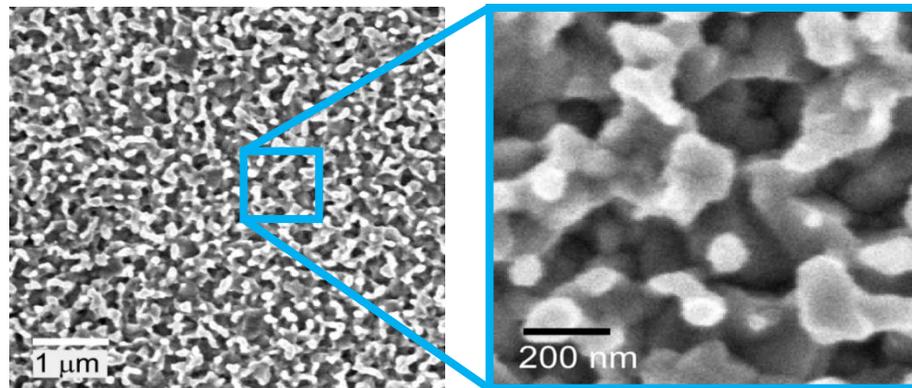
The films were placed in the tube furnace
at the room temperature and the temperature rate of heating and cooling was set 5C/min.

Sample 2

Sample 4

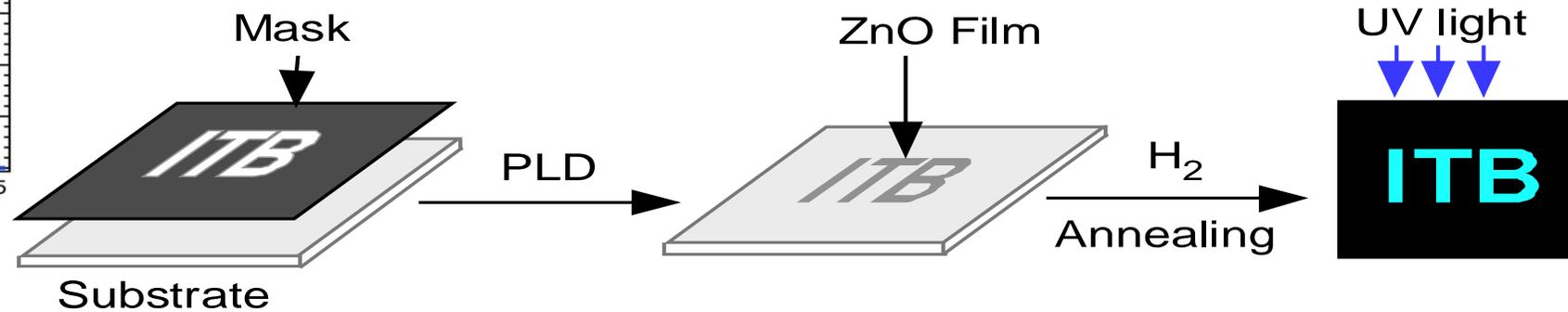
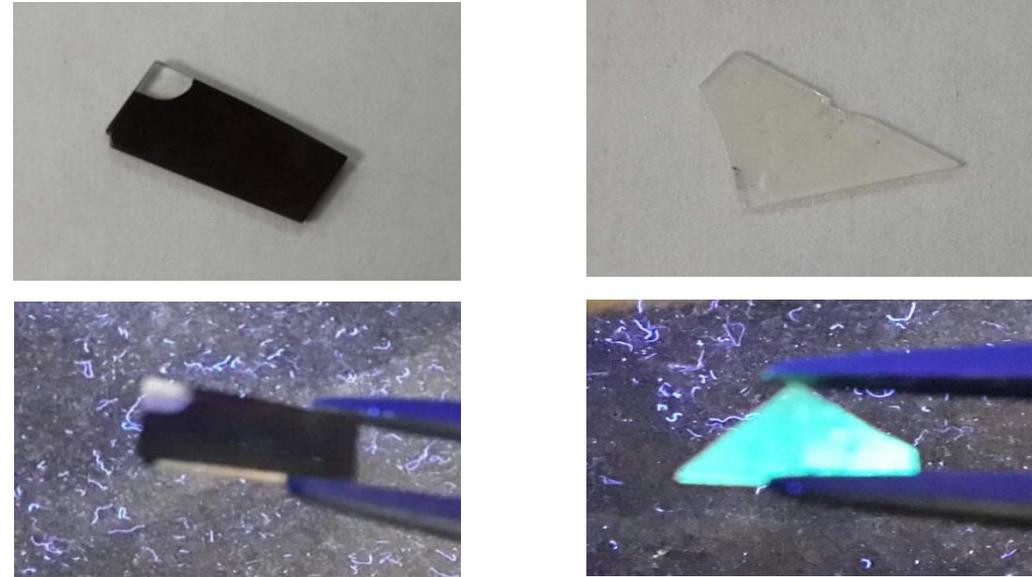
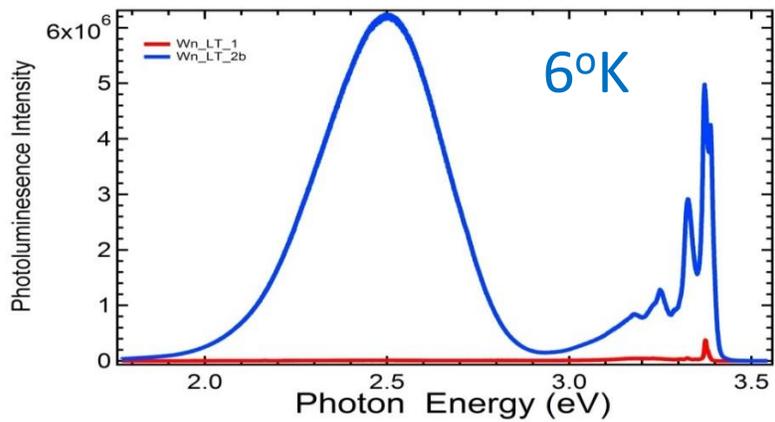
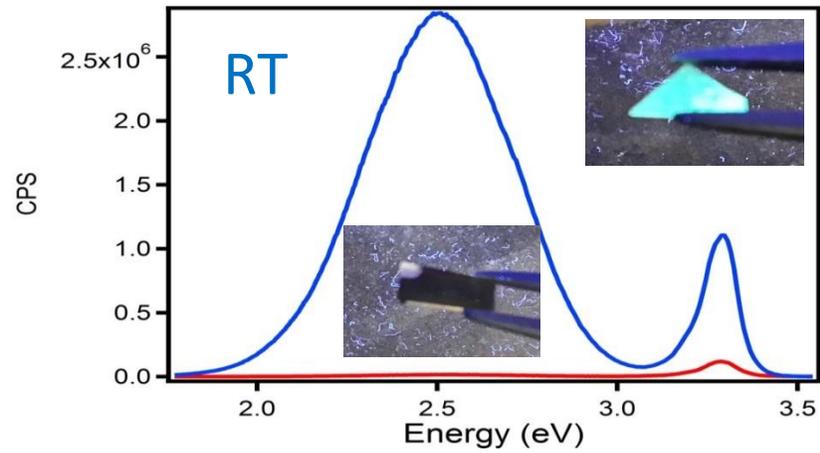


H2 Anneal
570 C , 1 Hour
~760nm



Photoluminescence (RT and 6K)

UV exposure



ZnO Film → ZnO Nanocolumnar (NC)

Sputtering → annealing

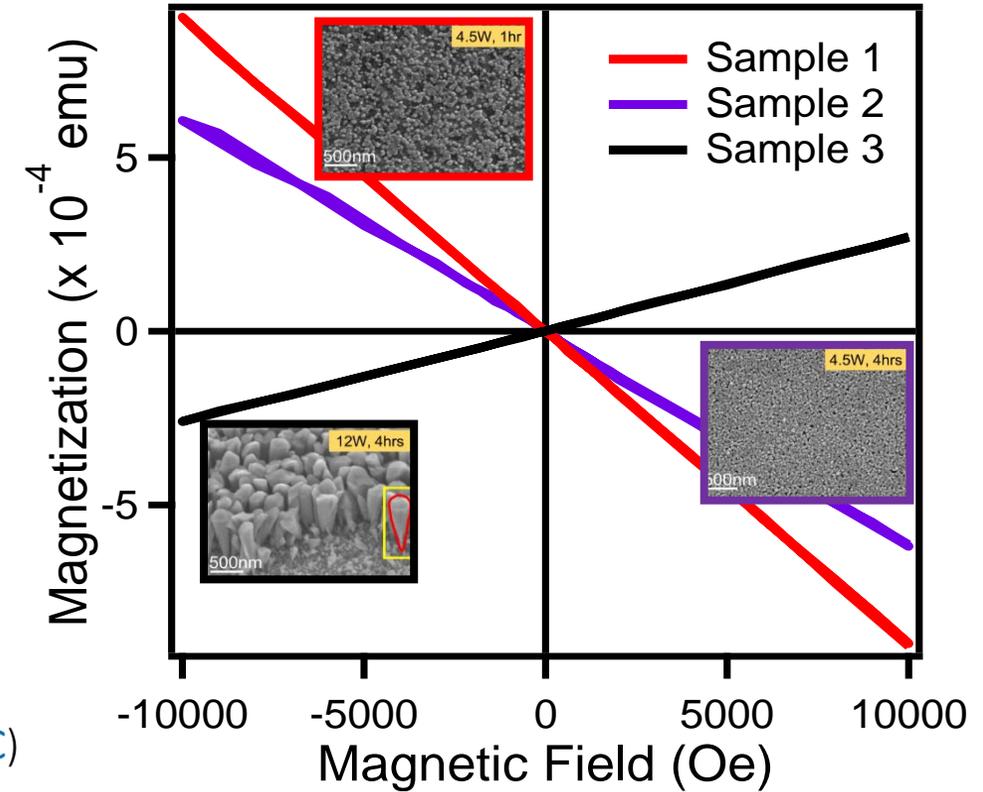
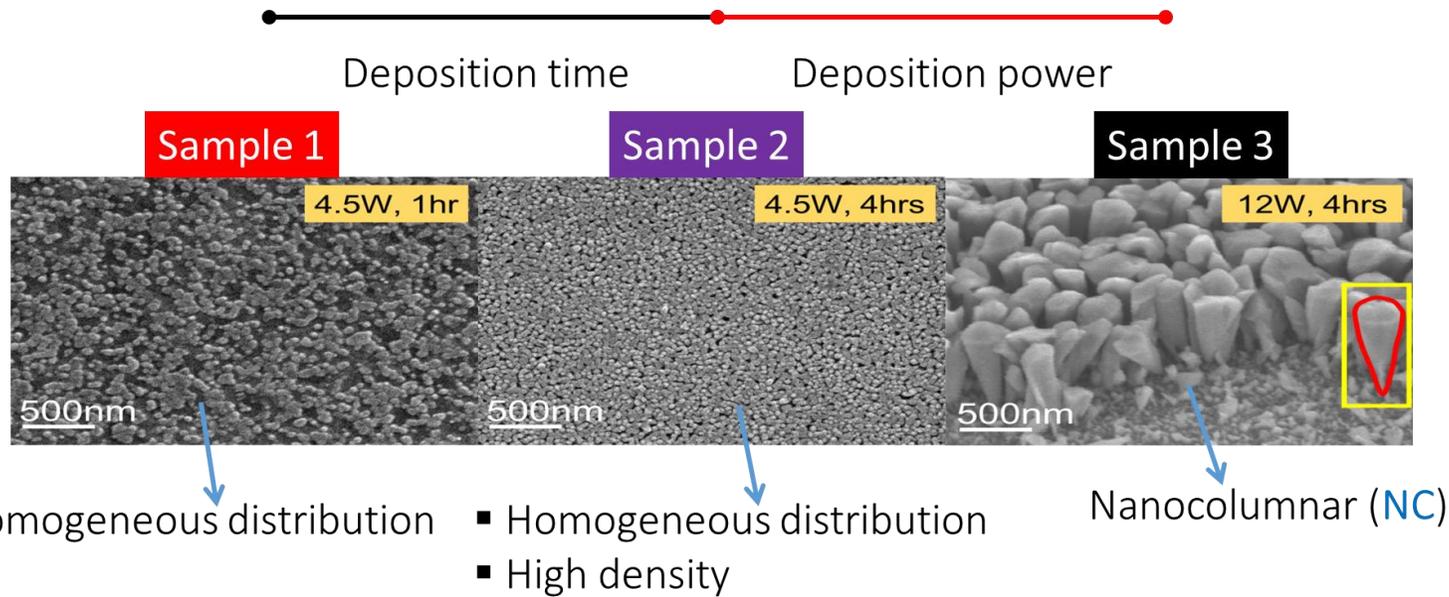
modification of magnetic and electronic properties of ZnO System

→ Reduction of ZnO dimension to increase Zn vacancy at the surface

→ Ferromagnetic-Semiconductor (NC)

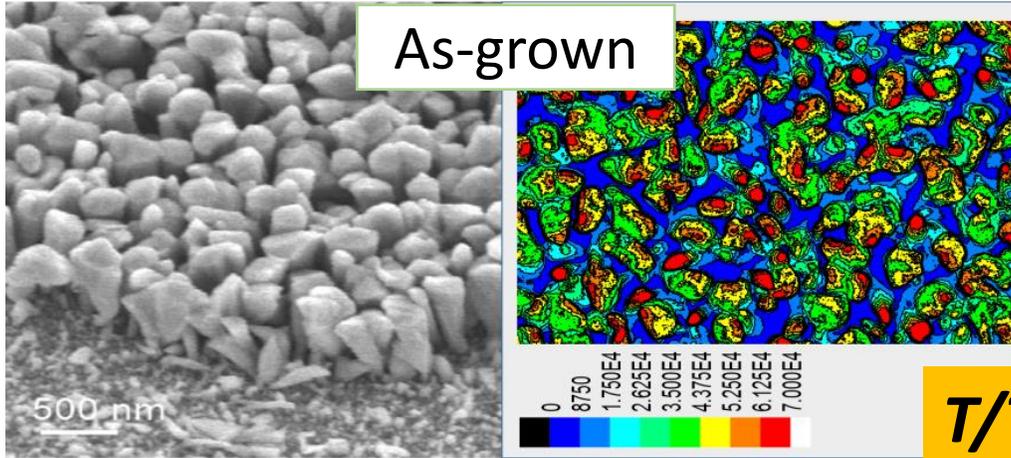
Magnetic properties of the ZnO

SEM images of ZnO under different deposition parameter



Sample 1 : diamagnetic
Sample 2 : diamagnetic
Sample 3 : paramagnetic

Effects of thermal annealing on the structure of the ZnO NC

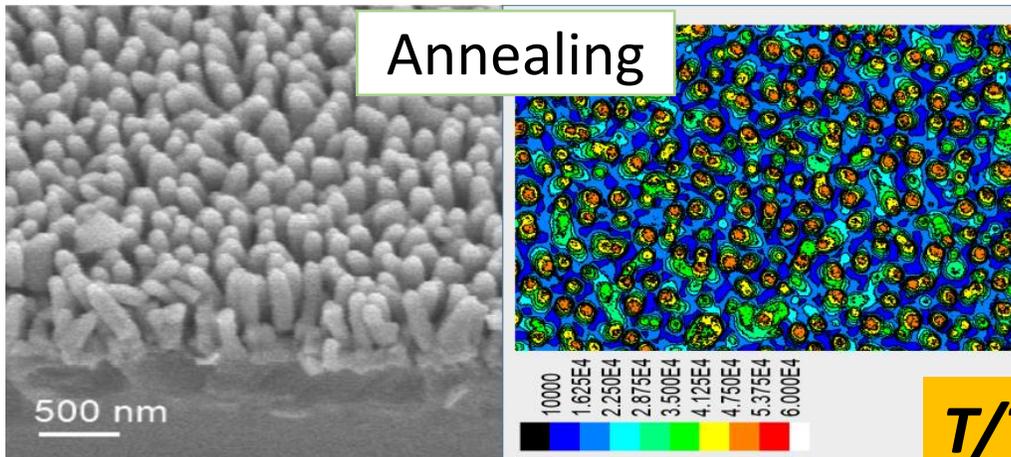


Non-uniformity of mass transfer during deposition

Promotes an inhomogeneous size along the vertical direction, forming a cone-like shape ($T/T_m \sim 0.2$ zone I).

$T/T_m \sim 0.2$

Post-annealing at 600°C + O₂

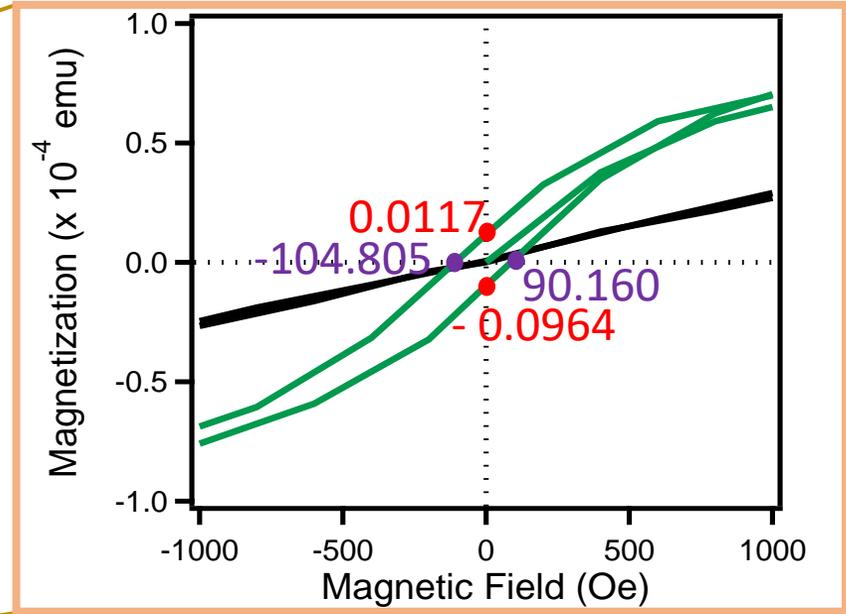
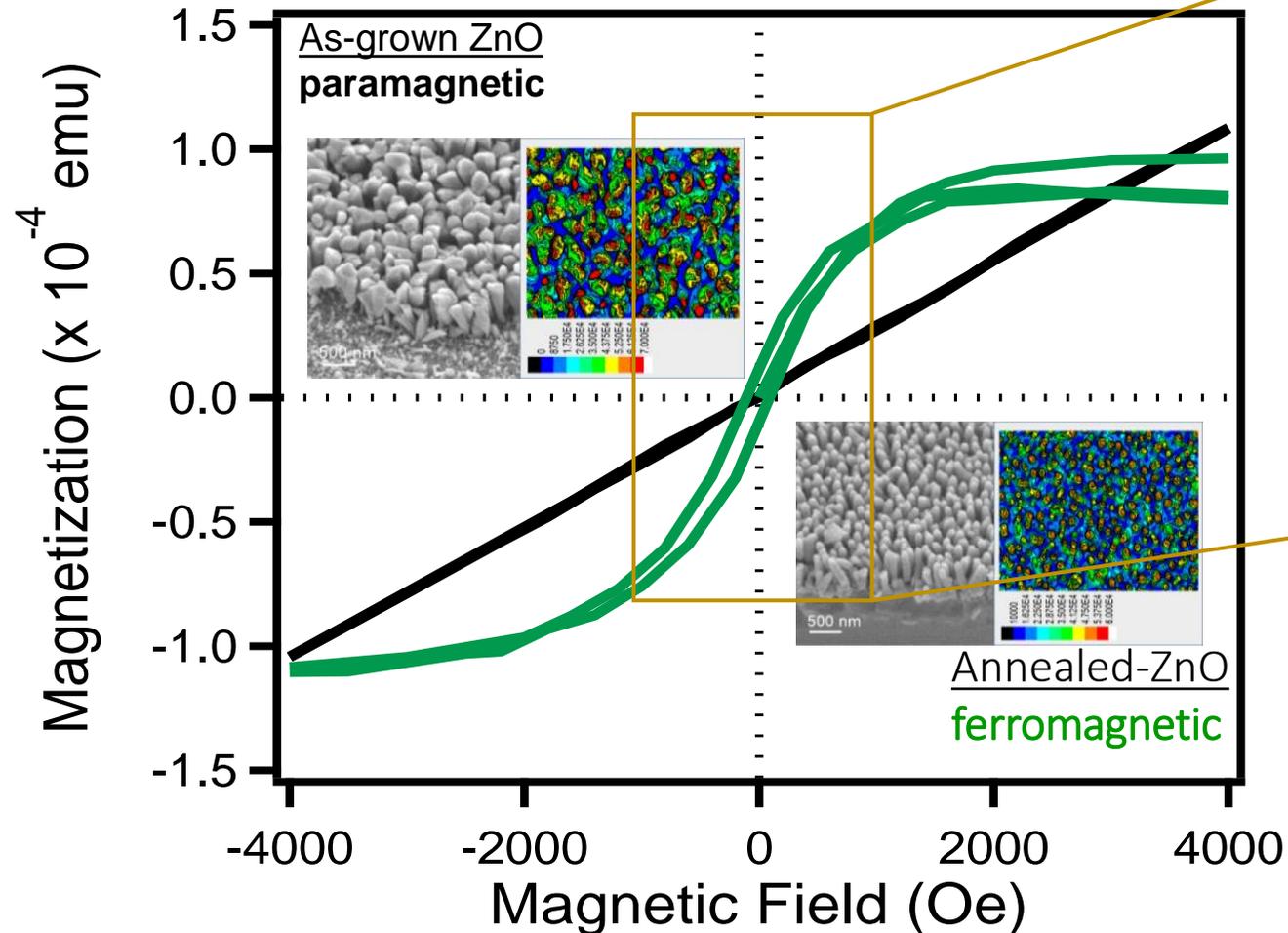


Post-annealing at 600°C promotes structural change ($T/T_m \sim 0.3$ zone II)

- Structure changes from cone-like to rods shapes.
- O₂ exposure promotes the decrease in thickness.

$T/T_m \sim 0.3$

Magnetization as a function of applied field



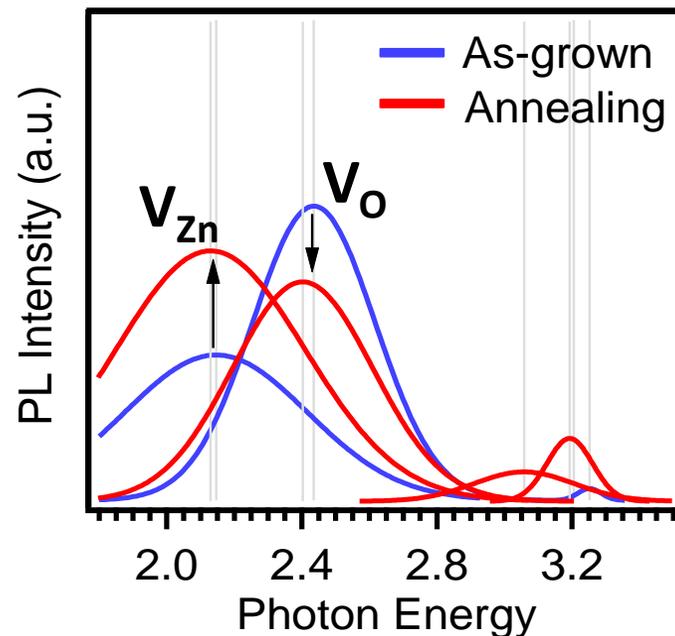
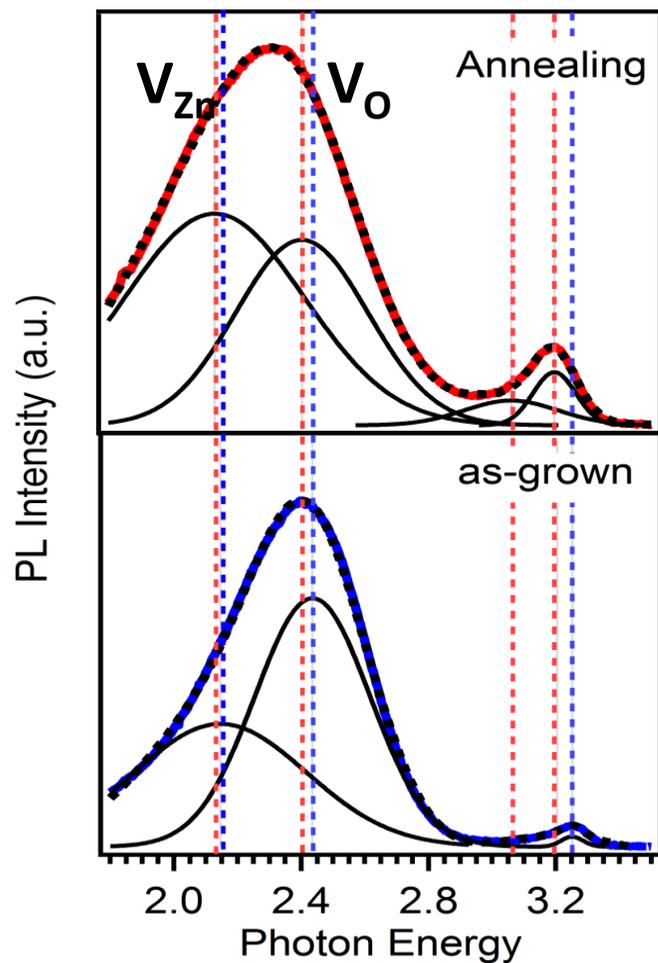
Properties	As-grown ZnO	Annealed-ZnO
$+M_r$ (emu)	-	1.17×10^{-6}
$-M_r$ (emu)	-	-9.64×10^{-6}
$+H_c$ (Oe)	-	90.160
$-H_c$ (Oe)	-	-104.805

□ As-grown ZnO
Linear line → paramagnetic

□ Annealed-ZnO
“S” profile → ferromagnetic

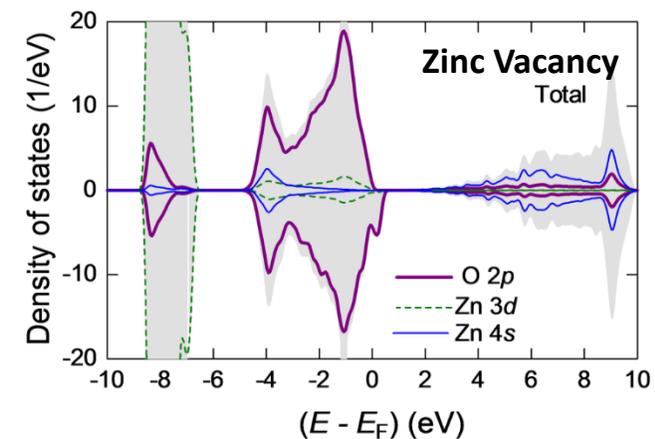
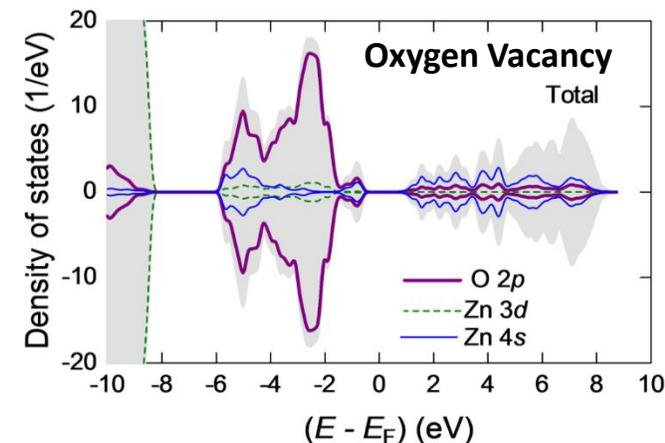


Photoluminescence spectra of the ZnO NCs



- As-grown ZnO
paramagnetic →
Domination of V_O defect
- Annealed-ZnO
ferromagnetic →
Domination of V_{Zn} defect

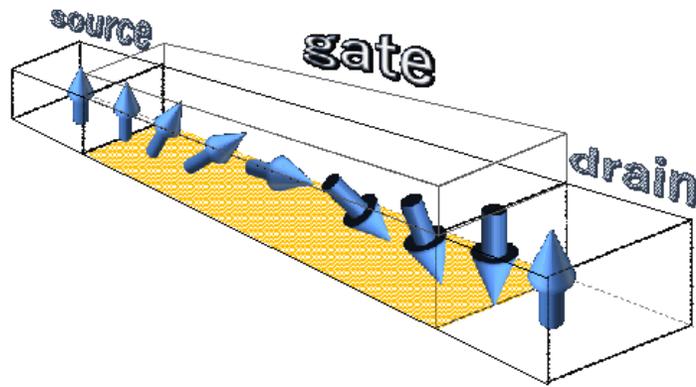
Density of States of the ZnO



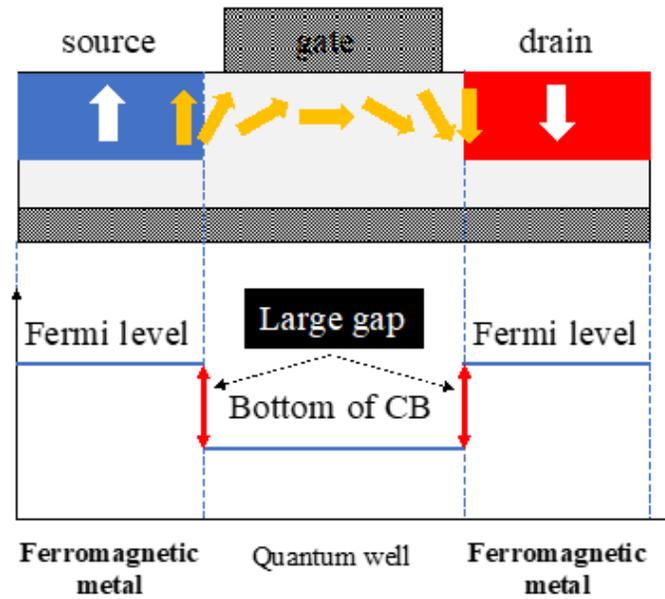
Zn vacancy (V_{Zn})-induced RT ferromagnetism

Ferromagnetic-Semiconductor ZnO for Spin FET

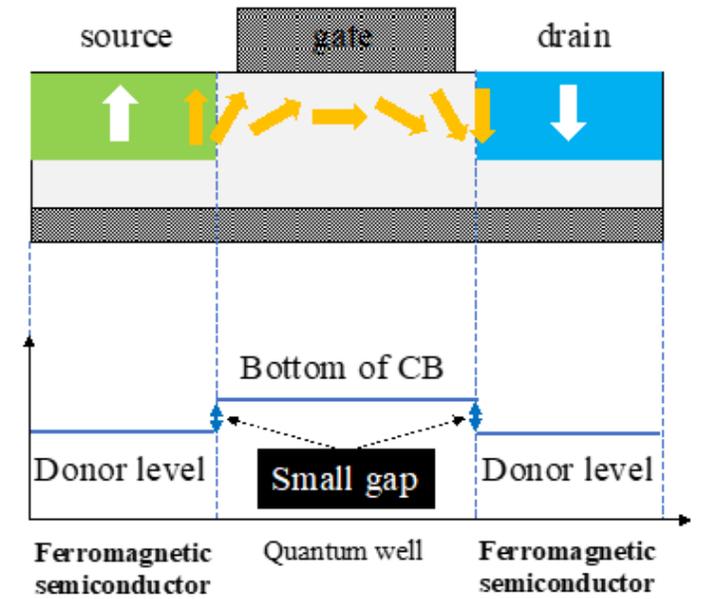
Spin Field Effect Transistor (FET)



Datta & Das type



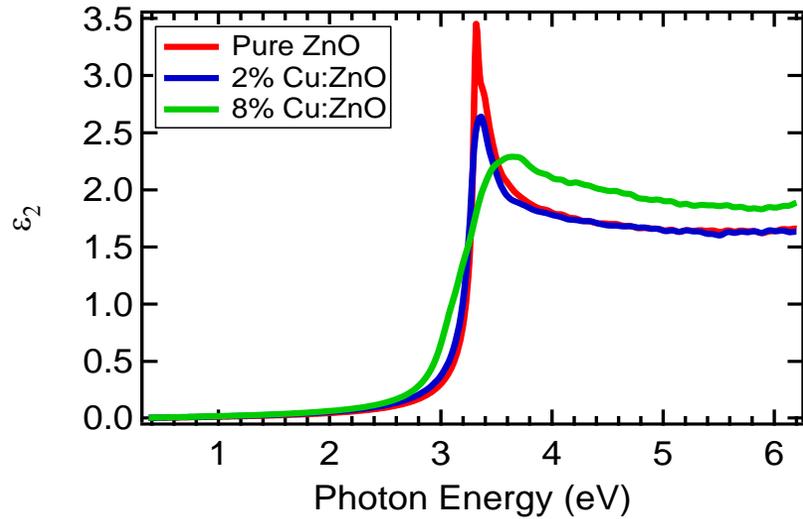
Proposed type



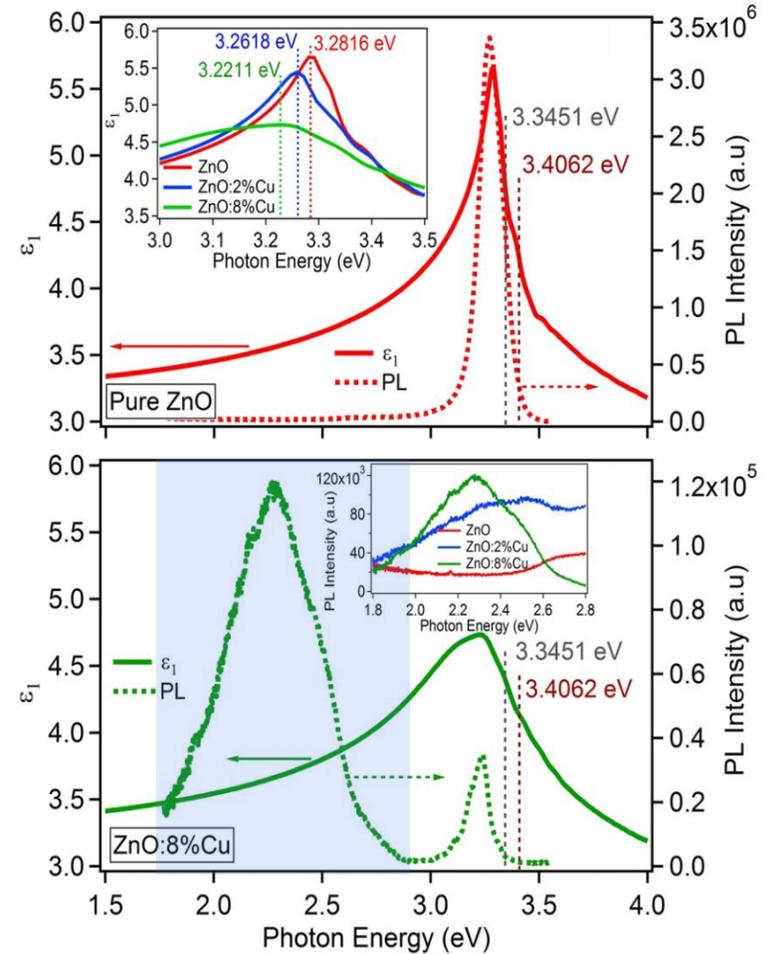
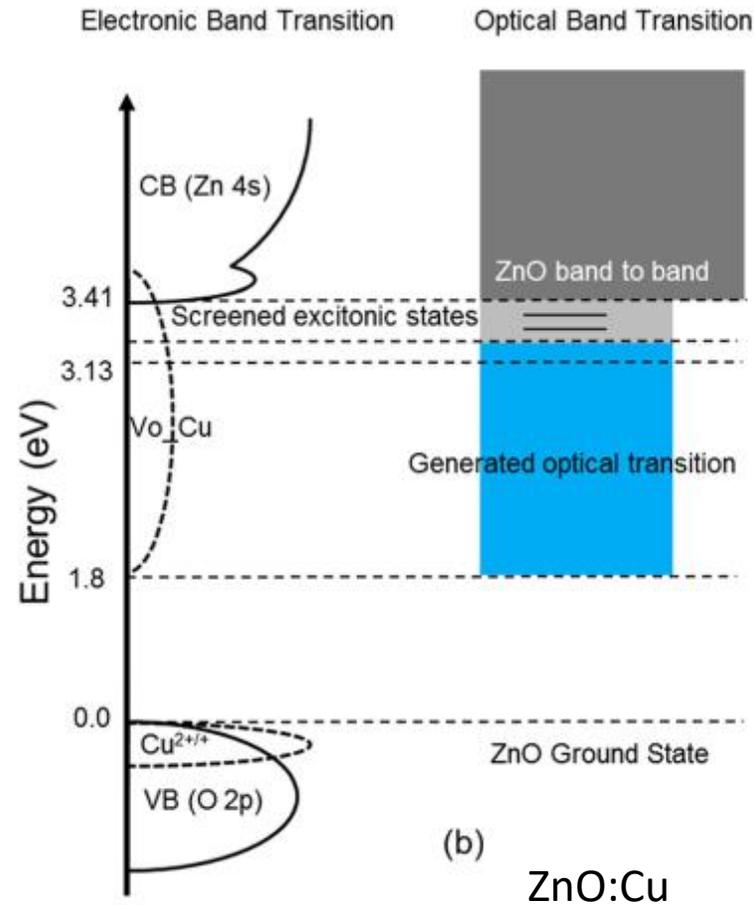
Appl. Phys. Lett. 56, 665, 1990

Ferromagnetic semiconductor is expected to play an important role as electrodes providing spin polarized electron in next generation device of spin FET

Doping; Exciton in ZnO:Cu



System with Cu impurities, a few quasi-free electrons may be provided by Cu 3d⁹. This leads to electronic screening effects and reduced the intensity of excitons.



Interplay of Cu and V_O
→ mid-gap optical transitions

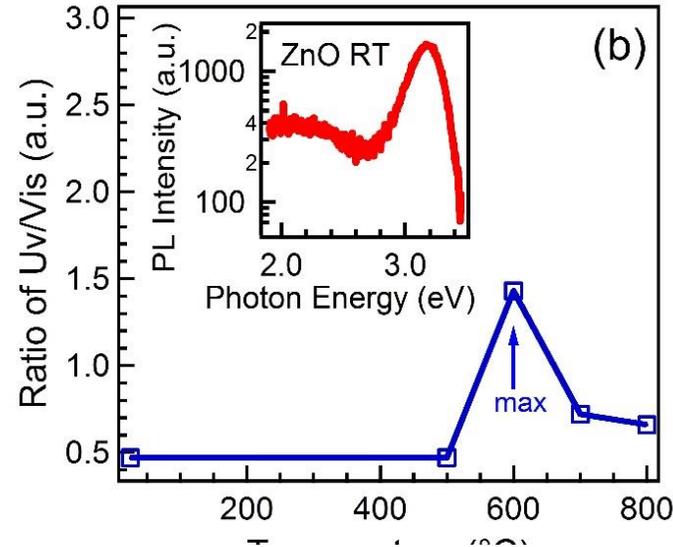
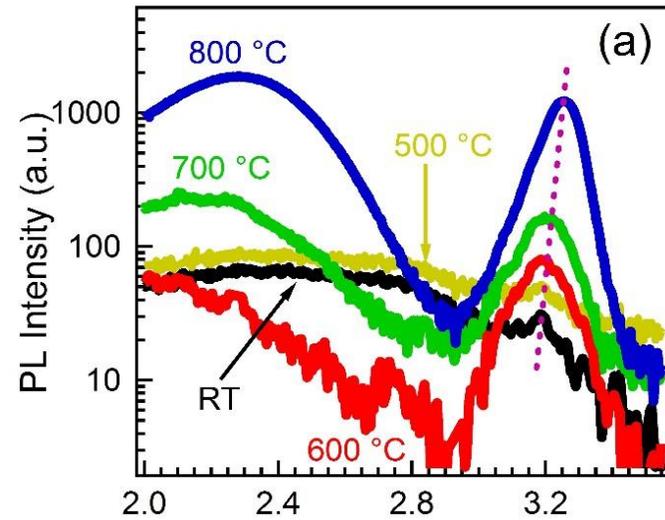
SE and PL techniques

→ powerful to detect defects bands and excitonic peaks

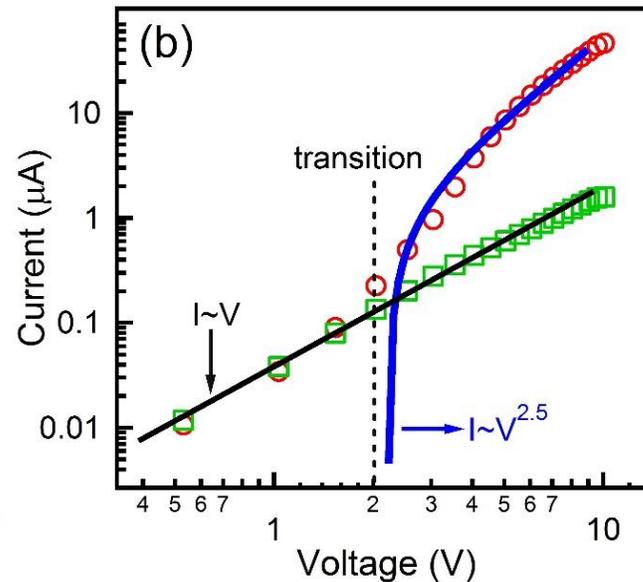
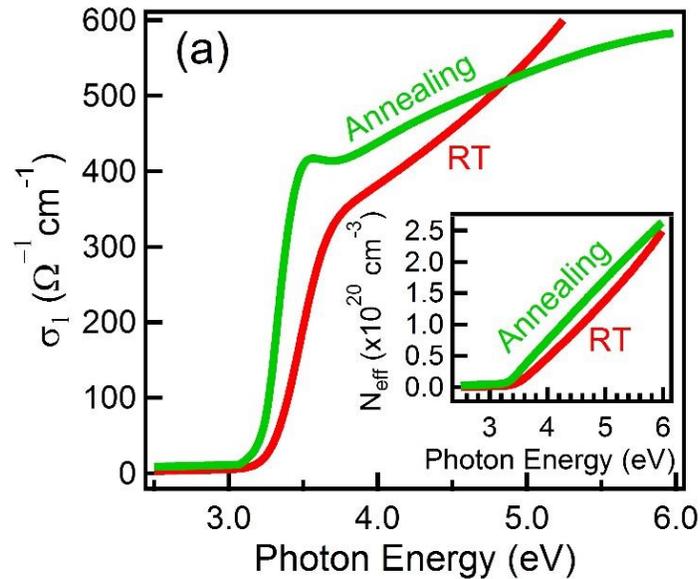
App. Phys. Lett. 104, 081922 (2014)



Doping; Effect of annealing temperature (ZnO:Ti)

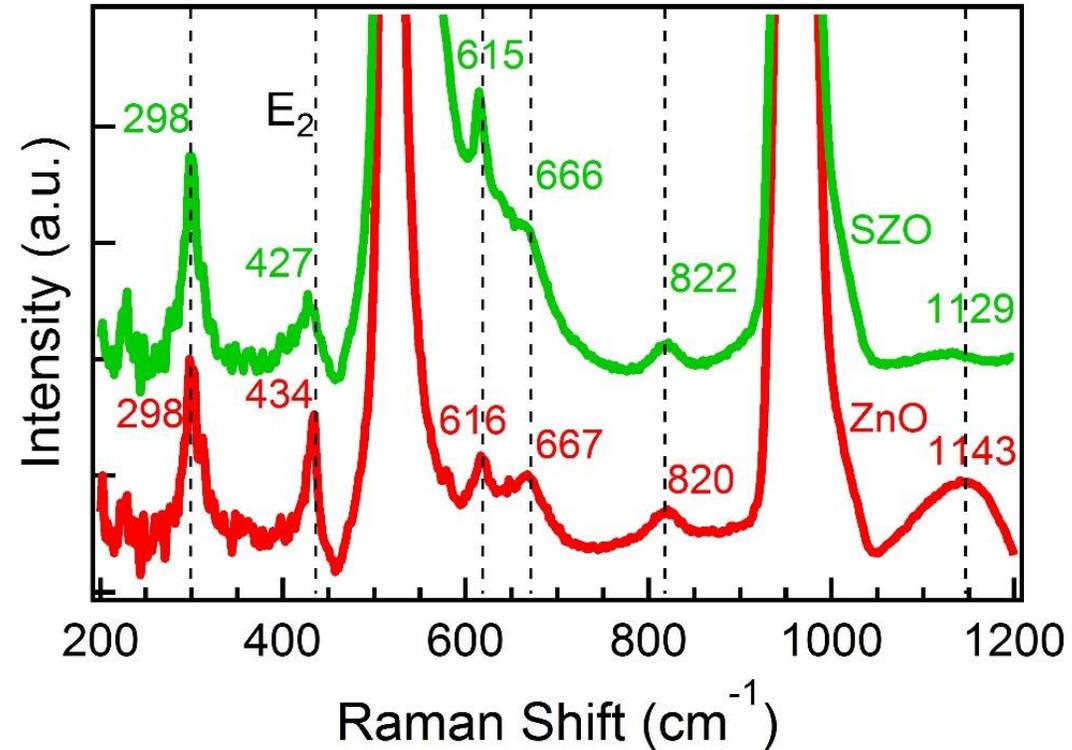
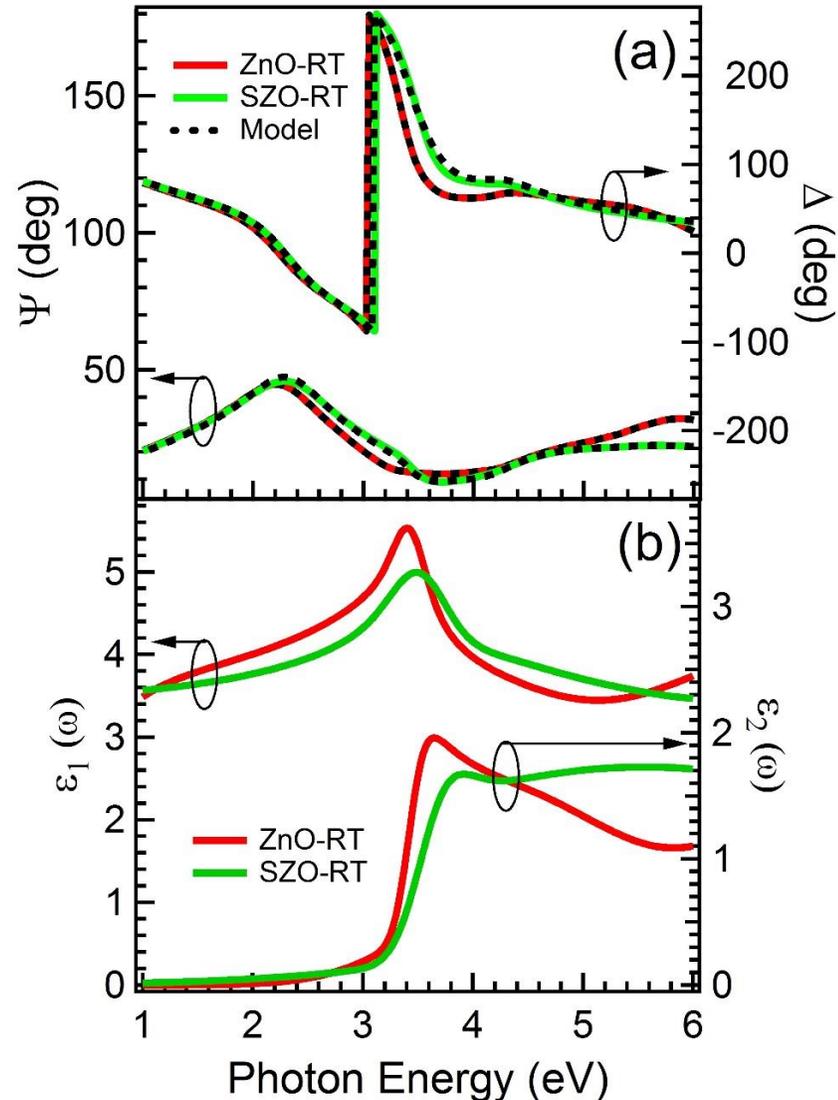


The contribution of oxygen vacancies
→ green emission (~2.5 eV)
→ electrical properties (space charge limited current regime)



Doping; Defects in ZnO:Sn Thin films

Sn dopant



Sn doping promotes the formation of oxygen vacancy-related trap states as indicated by A1 LO mode in Raman spectra and green emission in photoluminescence spectra.

Carbon-doped ZnO

properties



Optimized



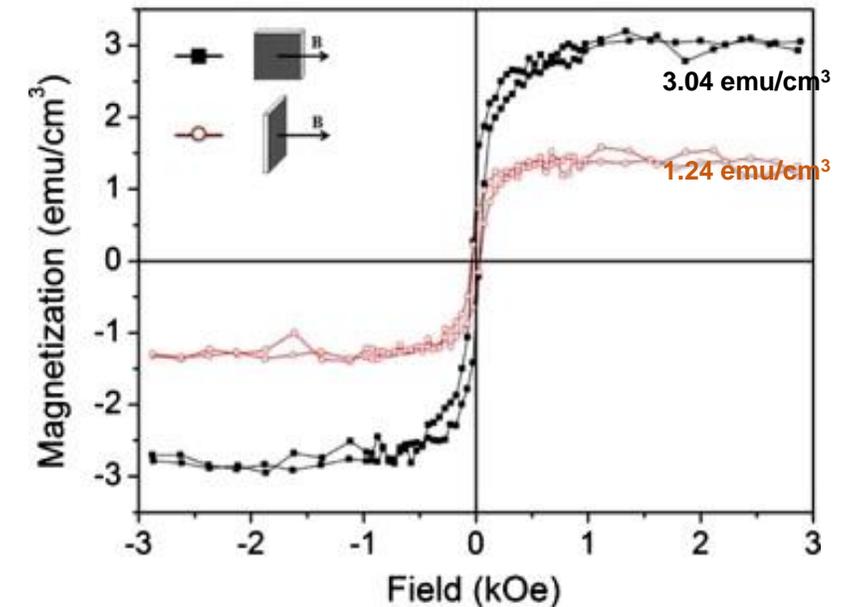
Dopant

Properties:

- room temperature ferromagnetic response
- hybridization of Zn and C ions
- an important role in polarization and optical characteristic

However, no report on the polarization characteristic of C-doped ZnO

The Ferromagnetism of ZnO:C



C.S.Wei., et al., Appl. Surface Science, Vol. 258, No. 14, 2012

→ investigate the polarization and optical behavior in C-doped ZnO nanocolumnar structure, which has high polarity.

Ex: Sample preparation and characterizations

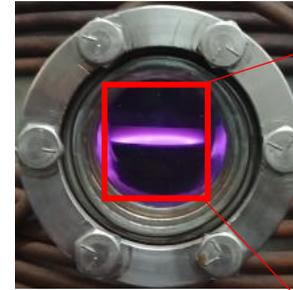
Sputtering target



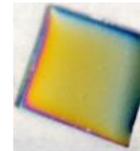
Substrate preparation



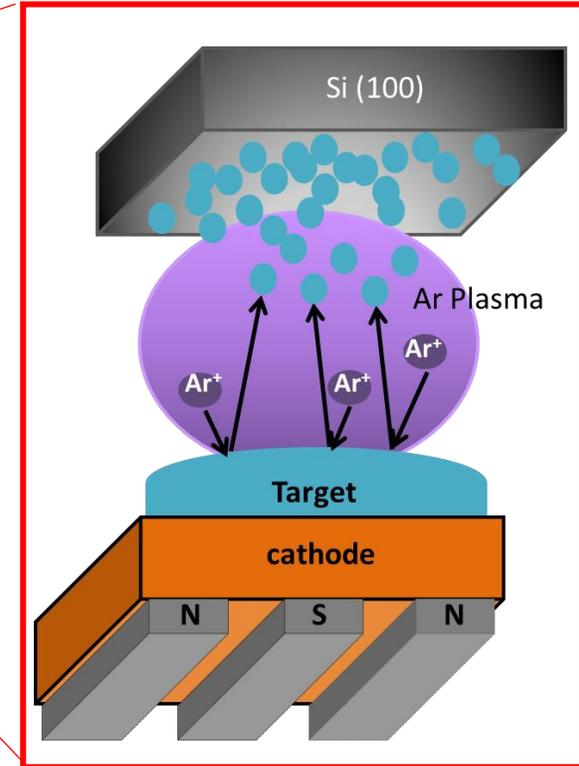
Deposition process :
DC-unbalance magnetron sputtering



Parameters;
 $t = 4$ hours,
 $T_{\text{substrat}} = 300$ °C
 $P = 0.3$ kPa,
Power = 12 W



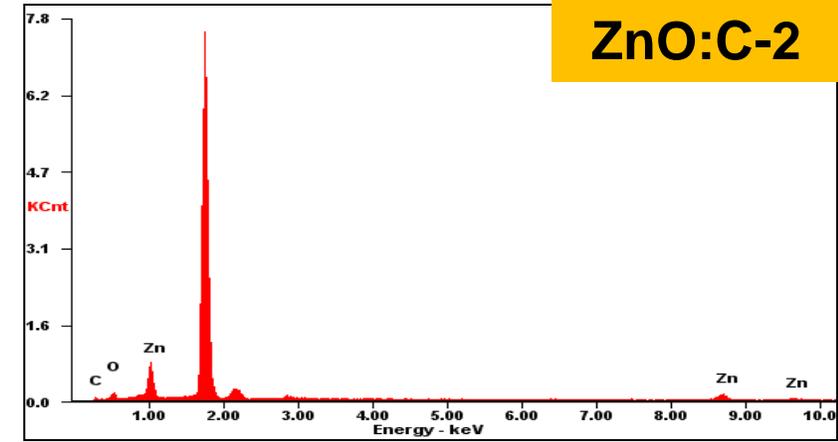
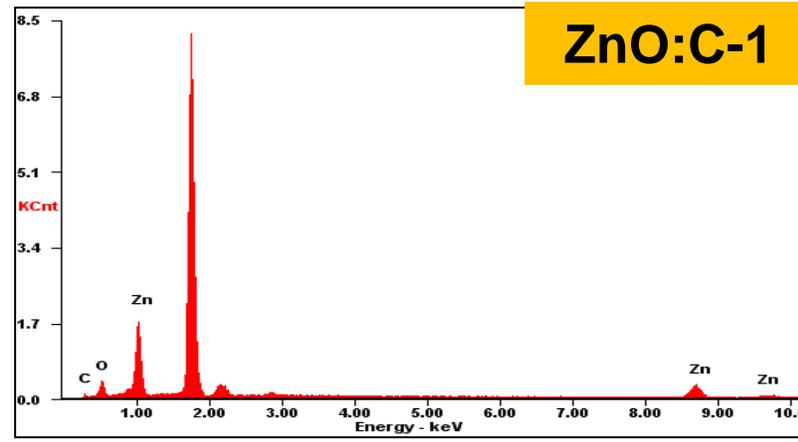
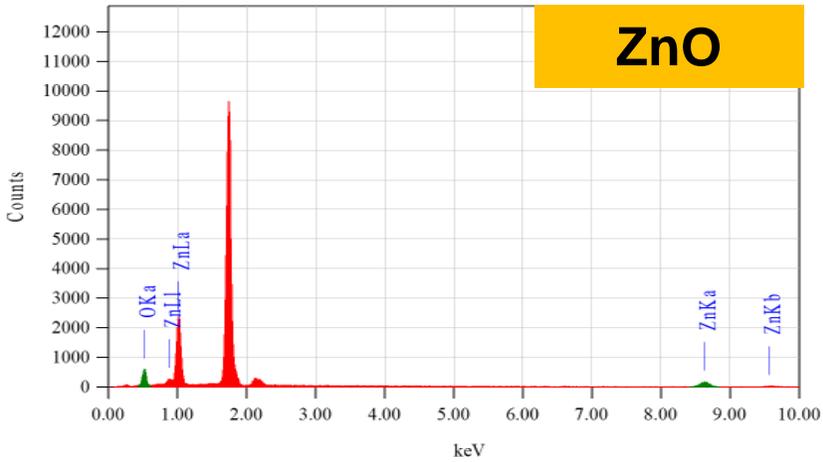
- ZnO
- ZnO:C-1
- ZnO:C-2



Characterizations:

FESEM, EDX, XRD, RT66A ferroelectric test system, PL, and UV-Vis.

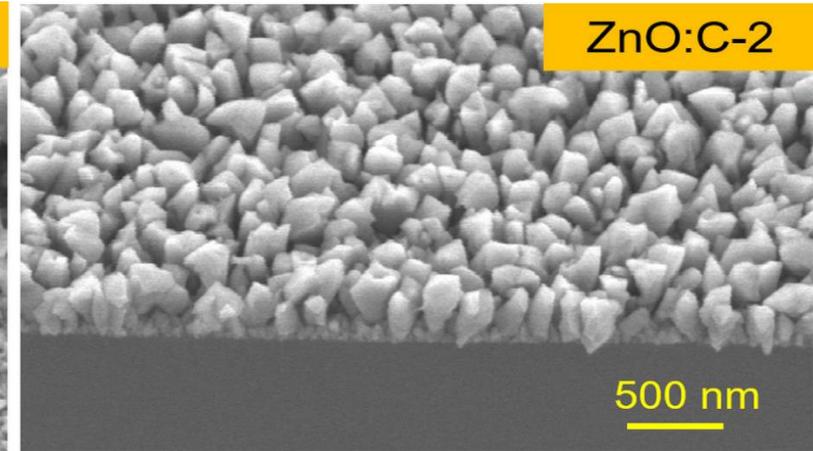
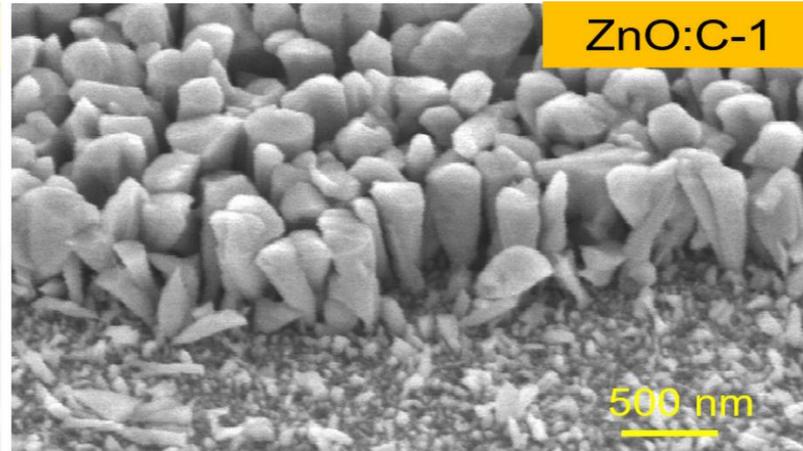
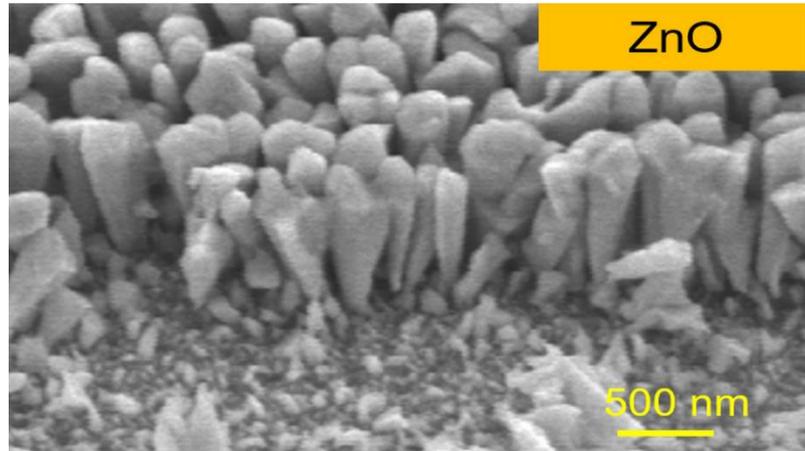
Results and Discussions: EDX Spectroscopy



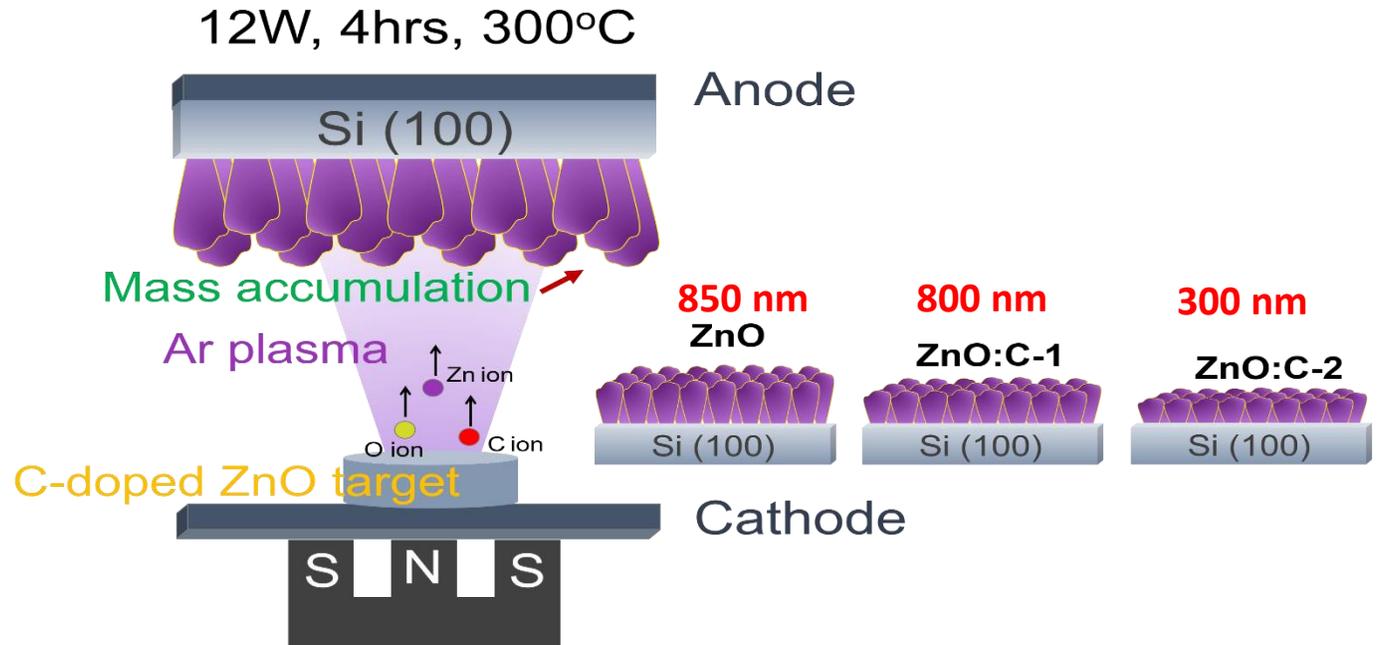
Sample name	Atomic concentration (%)			Total (%)
	Zn	O	C	
ZnO	36.86	63.14	0	100
ZnO:C-1	40.05	47.52	12.43	100
ZnO:C-2	30.38	37.86	31.76	100

The C atom is successfully doped with ZnO

Surface morphology



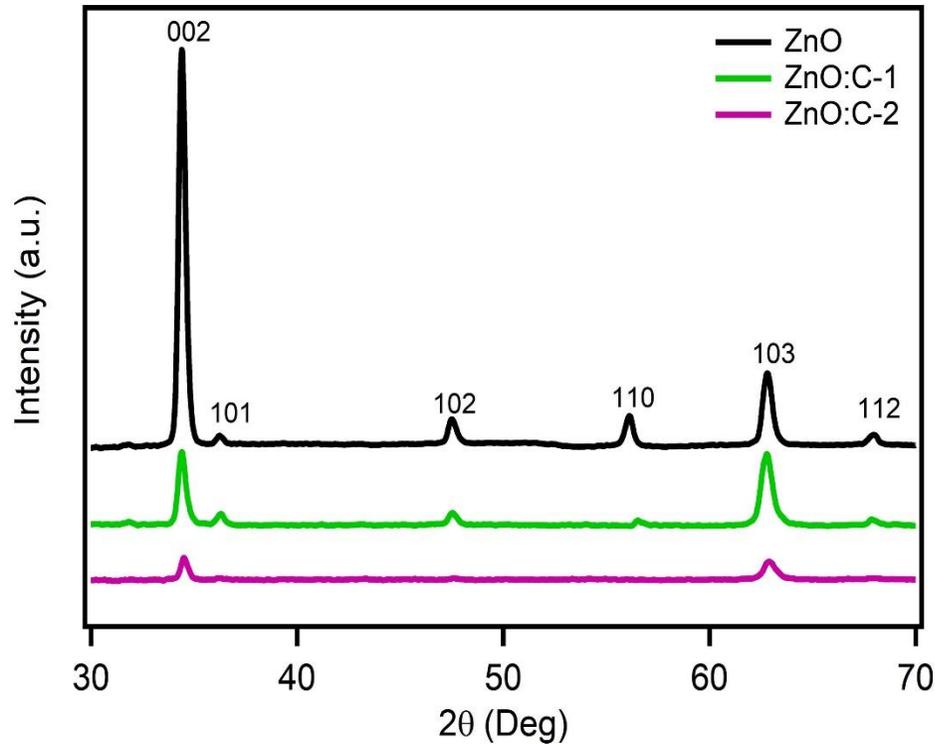
Hendri, et al., (2022). *Ceramics International*, 48(2), 2038-2044.



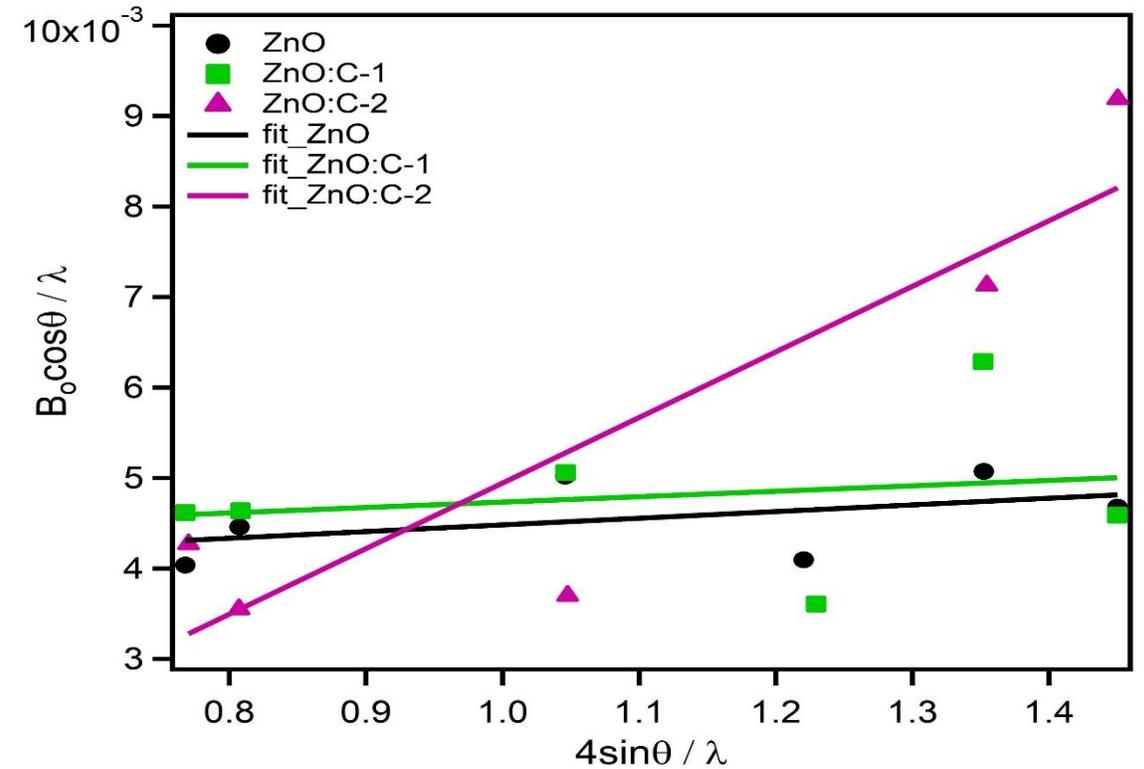
The smallest thickness of C-doped ZnO NCs by increasing the C concentration due to the different sizes of C, Zn, and O atoms will increase the chance of collisions between the atoms and affect the mean free path of the atoms.

Crystal structural

XRD pattern

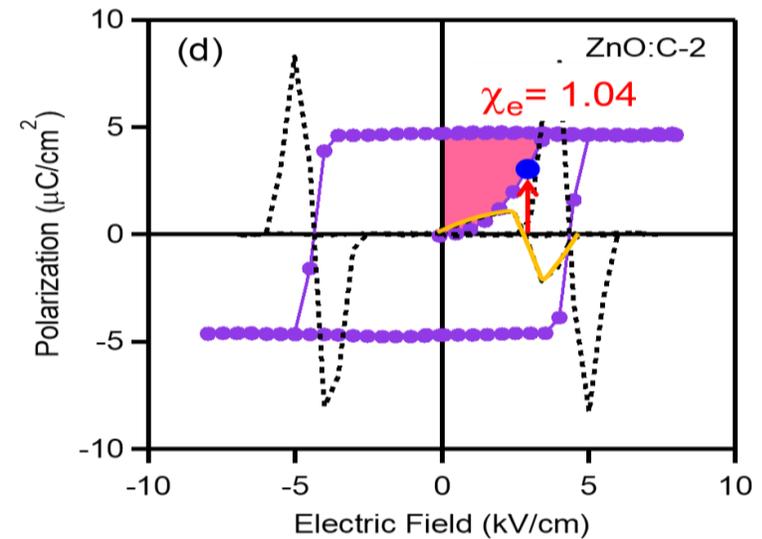
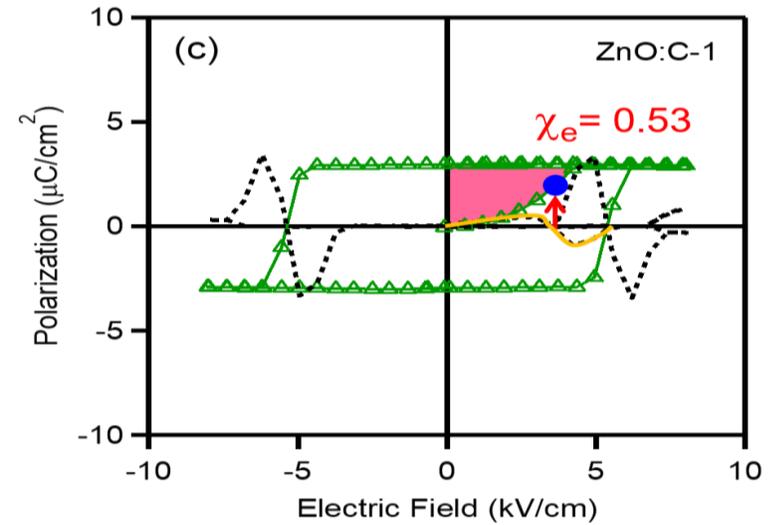
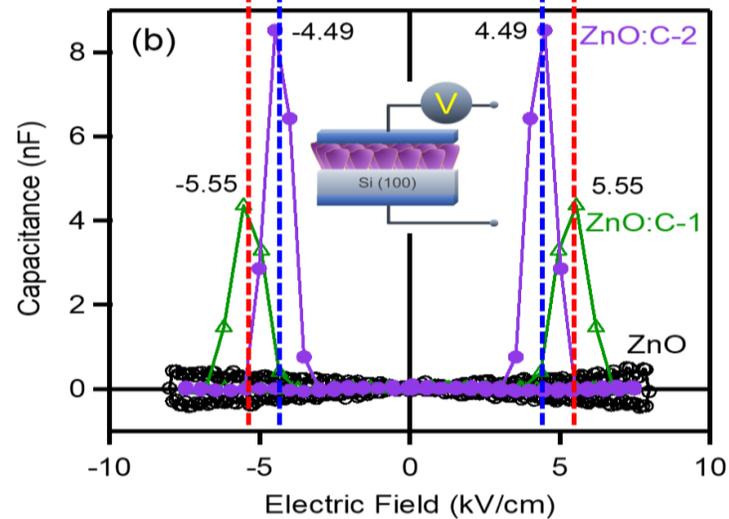
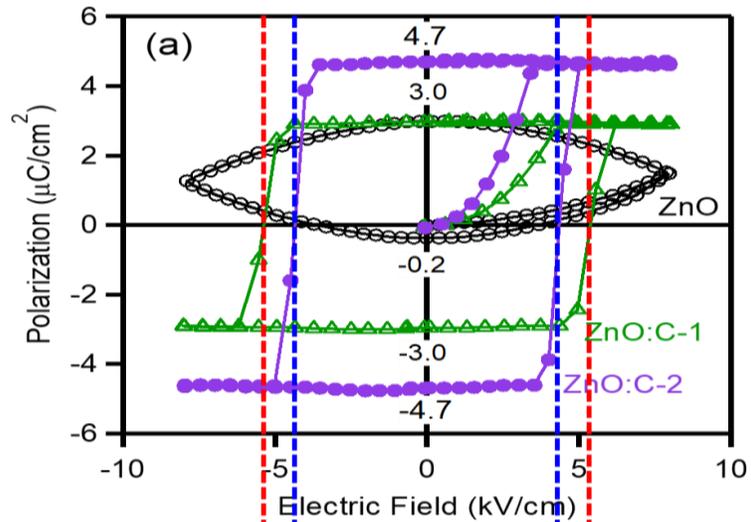


Williamson-Hall plot



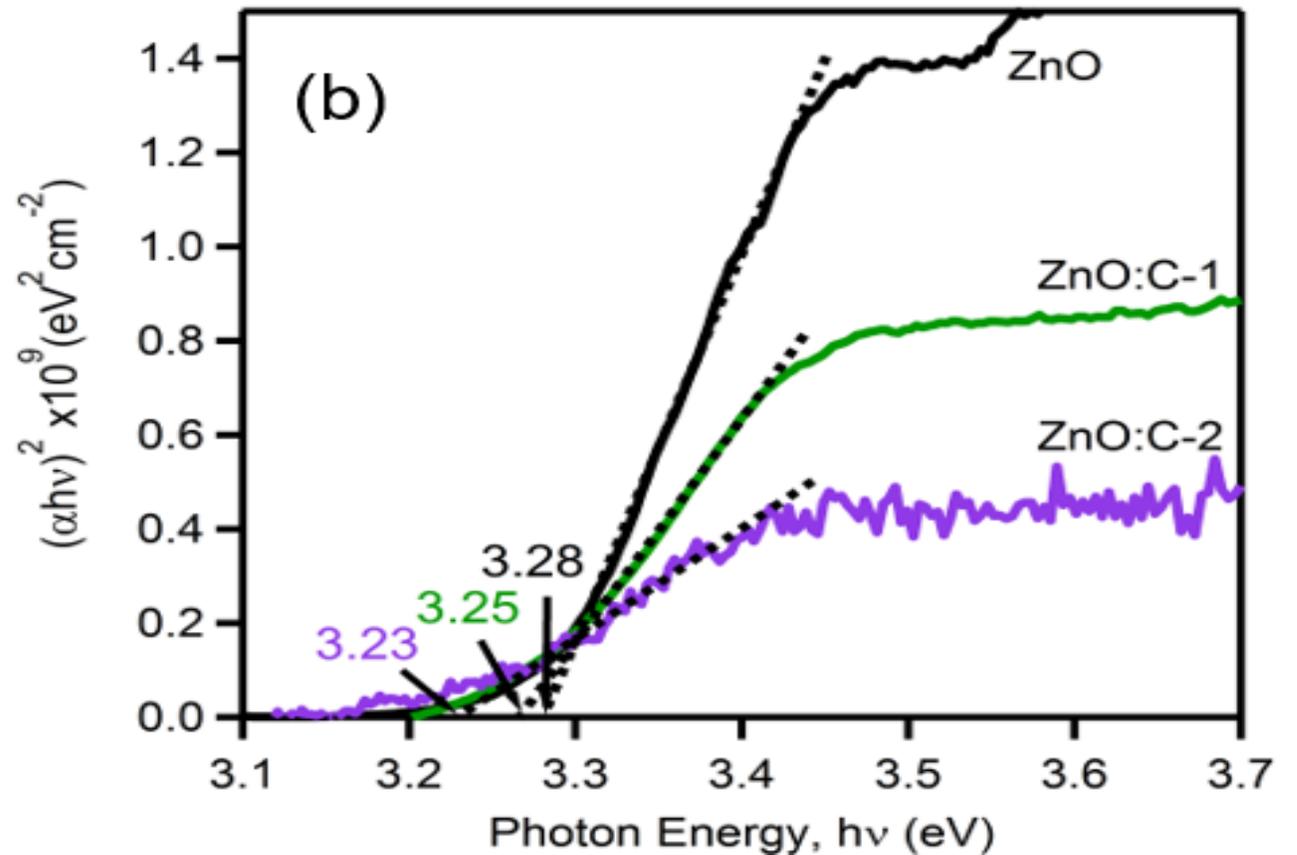
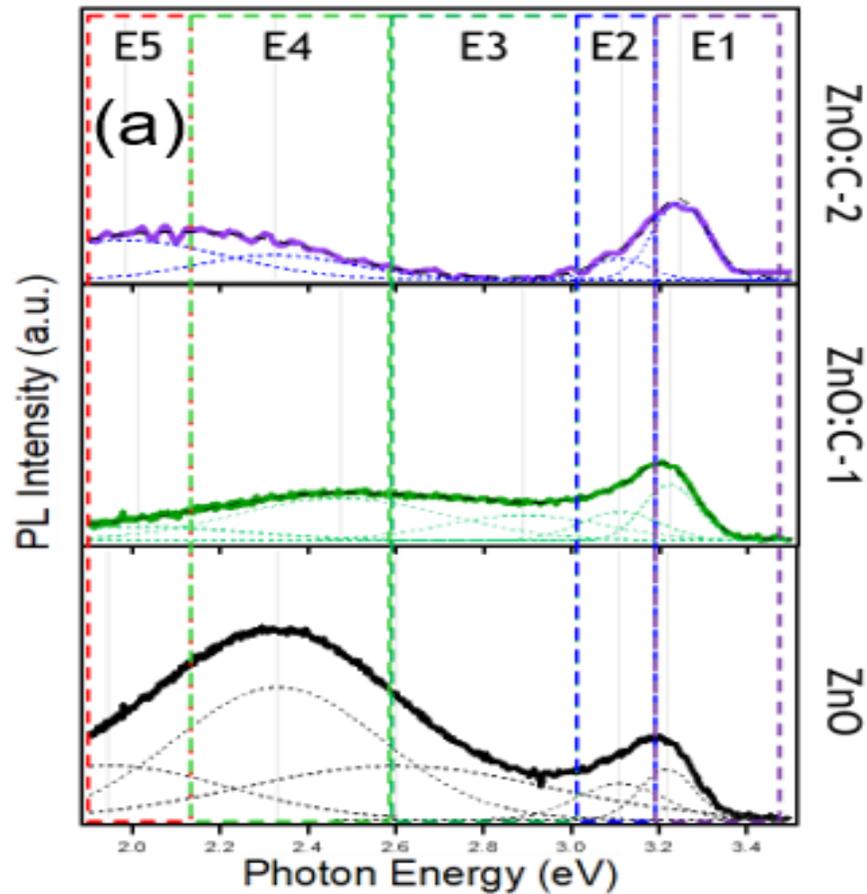
The increase of carbon concentration led to increase of lattice strain (ϵ) and decrease of crystallinity in dominant at (002) peak intensity.

Polarization and capacitance



The ferroelectric characteristics of the material are indicated by the butterfly capacitance curve with the sharp switching point, which indicates the change of polarity

Photoluminescence and absorption spectra



- ❑ The addition carbon in ZnO leads the increase in NBE emission accompanied by decreasing in defects emission
- ❑ The increase of carbon concentration promotes the decrease in absorption intensity accompanied by the red shift

List table of emission of the C-doped ZnO NCs

Sample name	Emission	Description
ZnO	E1 (3.219eV)	E1: near-band-edge (NBE) emission
	E2 (3.111eV)	E2: transition from Zn_i to valence band
	E3 (2.600eV)	E3: transition from Zn_i to V_{Zn}
	E4 (2.333eV)	E4: transition from V_O^* to valence band / V_{Zn} emission
	E5 (1.946eV)	E5: transition from conduction band to V_O^{++}
ZnO:C-1	E1 (3.222eV)	
	E2 (3.109eV)	
	E3 (2.890eV)	
	E4 (2.473eV)	
	E5 (2.015eV)	
ZnO:C-2	E1 (3.245eV)	
	E2 (3.112eV)	
	E4 (2.323eV)	
	E5 (1.986eV)	

Thank you

